Nanomaterials in construction – what is being used, and where?

Wendy Jones PhD, RN, SPCHN(OH), C. ErgHF
Researcher, School of Civil and Building Engineering, Loughborough University, Loughborough, UK (corresponding author: w.jones2@lboro.ac.uk)

Alistair Gibb PhD, CEng, MICE, MCIOB
ECI Royal Academy of Engineering Professor, School of Civil and Building Engineering, Loughborough University, Loughborough, UK

Chris Goodier PhD, MCIOB, FICT, FHEA
Senior Lecturer, School of Civil and Building Engineering, Loughborough University, Loughborough, UK

Jie Jin PhD
Researcher, Department of Materials, Loughborough University, Loughborough, UK

Phil Bust PhD
Researcher, School of Civil and Building Engineering, Loughborough University, Loughborough, UK

Mo Song PhD, MCSA, MCCS
Professor in Polymer Science and Nanotechnology, Department of Materials, Loughborough University, Loughborough, UK

1. Introduction

Nanoscale refers to dimensions in the range 1–100 nm, with nanomaterials being those that contain particles (or sometimes spaces or pores) in this range in at least one plane. To appreciate how small nanoscale is, consider that a human hair is around 100 000 nm thick (0.1 mm). Figure 1 illustrates this in another way: the ratio between a carbon fullerene and a small grapefruit is about the same as the ratio between the grapefruit and the Earth.

Nanomaterials can be found in foods, textiles and cosmetics as well being used in the electronics, automotive and aeronautical industries (Keller et al., 2013). Nanotitanium and nanozinc are added to sun protection creams (Schilling et al., 2010); nanosilica is used as a filler and whitener in toothpastes; and nanosilver is added to socks and other consumer goods for its anti-microbial properties, as well as being used in medical and food hygiene applications (EC, 2012).

Nanomaterials are also being used increasingly in construction (AECOM, 2014; Teizer et al., 2009) and offer many useful and desirable properties. Their addition to concrete, for example, can improve compressive and tensile strength, and abrasion resistance. Associated with this is the (claimed) potential to reduce global cement consumption, with the associated energy and environmental benefits that this entails (Raki et al., 2010). Energy savings may also come from the use of very effective nano-enabled insulation materials; and nanomaterials could replace more toxic compounds in coatings (Pagliaro et al., 2009). The use of nanomaterials is predicted to rise substantially, accounting for 50% of building materials by the year 2026 (AECOM, 2014).

Although there is much published about nanotechnology and nanomaterials in construction (over 300 papers in Construction Materials alone in the last 10 years), it can be difficult to obtain a clear picture of exactly which materials have been used, where, and in what quantities. Not all materials that show promise have been, or will be, successfully converted into marketable products – there is, for example, a substantial difference (practically and financially) between laboratory research on small samples of cement paste and the realities of...
Nanomaterial definitions vary widely (David et al., 2013) – a particular substance may be classed as a nanomaterial in one country but not in another. This is a contentious area. The EU has been debating its definition of nanomaterials for over 5 years, with a revised version due imminently (Rauscher et al., 2013; Oberdörster et al., 2007; van Broekhuizen et al., 2012); a report by the International Agency for Research on Cancer (IARC, 2010) considers nanoscale titanium as being 1–150 nm in size; and materials with dimensions as low as 0·1 nm are also technically nanoscale (Zhu et al., 2004).

Some definitions, such as that of the EU (EC, 2011; ECJRC, 2014), only include materials that contain nanoparticles; others include nanostructured materials that have nanosized spaces or pores but no particles (e.g. BSI, 2011). Some definitions only include materials that are intentionally produced (often known as engineered nanomaterials or manufactured nanomaterials (MNMs); others include naturally occurring materials.

There is also variation in how definitions consider particles that are aggregated or agglomerated (bound together), what proportion of particles must be nanoscale for the definition to apply, and whether particularly novel properties must be present for categorisation as a nanomaterial.

These differences make it difficult to frame and implement regulations and to interpret manufacturers’ literature. They also create challenges when studying or writing academic papers, as different conceptualisations might be used in each case.

2.2 ‘Nanomaterials are new’

Some nanomaterials have indeed been created or discovered in recent years – examples include graphene, which is a layer of carbon an atom thick, isolated in 2004 (Novoselov et al., 2004), and carbon nanotubes (CNTs) (Iijima, 1991), essentially such a layer rolled into a tube. However, substances of nanoscale proportions have been around for much longer. Naturally occurring nanoparticles include some viruses and parts of the ash released by volcanos; naturally occurring nanostructures contribute to the colours of butterfly wings (Giraldo, 2008) and to leaf surfaces that are able to shed water easily (Grebler et al., 2010). Many coloured pigments used in paints and coatings are nanoscale, and have been so since the earliest cave paintings (Orfescu, 2014); steel-making has been based on nanotechnological methods for over 200 years (Kolpakov et al., 2007).

Although many novel nanomaterials are now being developed, the majority of nano-enabled products are still based on older definition at all. Variations between nanomaterial definitions include the following.

- Most definitions consider nanoscale as being 1–100 nm, but may include larger dimensions as long as other criteria are met. Some nanotoxicologists consider the upper boundary unhelpful as toxicology does not change abruptly at the 100 nm cut-off point (Donaldson and Poland, 2013; Oberdörster et al., 2007; van Broekhuizen et al., 2012); a report by the International Agency for Research on Cancer (IARC, 2010) considers nanoscale titanium as being 1–150 nm in size; and materials with dimensions as low as 0·1 nm are also technically nanoscale (Zhu et al., 2004).

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This research is part of a 3-year, multi-disciplinary project funded by the UK’s Institution of Occupational Safety and Health to explore the application and management of nanomaterials in the construction and demolition sectors through academic and commercial literature and interviews with stakeholders. A second objective was to test samples of nano-enabled construction products and establish the likelihood of nanoparticle release when buildings are demolished and the materials recycled. The results of these tests, plus additional details relating to the potential health risks of nanomaterials are reported elsewhere (e.g. Jones et al., 2015a, 2015b) (other papers are in progress).

2. Myths, misunderstandings and misconceptions

2.1 ‘Nanomaterials are clearly defined’

Nanomaterial definitions vary widely (David et al., 2013) – a particular substance may be classed as a nanomaterial in one country but not in another. This is a contentious area. The EU has been debating its definition of nanomaterials for over 5 years, with a revised version due imminently (Rauscher and Roebben, 2015); and the USA does not have a formal

Figure 1. A carbon fullerene, a grapefruit and the Earth
substances. For example, rubber for tyres contains both carbon black and synthetic amorphous silica (SAS) with SAS also being used as a filler or flow agent in powders, foods and toothpaste. Together, these two materials account for 97% of the nanomaterial use in Europe (EC, 2012; IARC, 2010).

2.3 ‘Nanomaterials are hazardous’

One group of nanomaterials that has prompted concerns regarding toxicity is CNTs, with newspaper headlines such as ‘The micro-particles that could pose the same risk as asbestos’ (Derbyshire, 2008). CNTs, like asbestos, are fibre-shaped particles, with a high aspect ratio (i.e. their length is at least three times their width, and often many times more). They are therefore able to penetrate deep into the lungs, but are difficult for the normal protective mechanisms to clear (Donaldson et al., 2013), and one specific type of CNT has been identified as carcinogenic by the IARC (Grosse et al., 2014). However, CNTs exist in many different forms (perhaps in the order of 50 000 different variants (WHO, 2012)) and they vary in their toxicity, with short or tangled forms appearing to be less toxic than those that are long and stiff (Donaldson et al., 2013). This could enable manufacturers to select nanomaterials that have safer dimensions (if they choose to, and provided this does not conflict with the required functionality). Other modifications such as small changes to the chemistry of CNTs, the use of coatings or different processing methods can also reduce their toxicity, so that CNTs can be designed that are intrinsically less hazardous (Al-Jamal et al., 2012; Costa, 2014).

The same variability exists for other nanomaterials. Nanomaterials have high surface area per unit of mass, making them more reactive and potentially contributing to increased toxicity (Oberdörster et al., 2007). However, the chemistry, shape, electrical charge, solubility and structure of materials will also influence their toxicity. For example, amorphous silica (the type typically used in nano-enabled construction materials) is much less toxic than crystalline silica, which, in its respirable, non-nano form, will be familiar to those working in construction as the cause of silicosis.

Due to the wide variability of nanomaterials, it is often difficult to compare the studies done to assess their toxicity, particularly as the majority of studies do not adequately describe the material used in terms of the characteristics mentioned above (size, shape, charge etc.) (Boverhof and David, 2010; Krug, 2014). It would be ideal if the nanomaterials tested in terms of their toxicity were identical to those used in products, but most testing is conducted with materials produced specifically for research purposes rather than on those taken from commercially available products. At this stage, therefore, it is only possible to state that some nanomaterials are hazardous, and that some are more hazardous than others.

2.4 ‘Nanomaterials are easy to identify’

There is no requirement under European or UK law for most nano-enabled products to be identified as such. Products labelled as ‘nano’ may contain nanoparticles, be nanostructured (i.e. they have nanoscale pores but no nanoparticles), be developed or manufactured using nanotechnology (but contain no nanomaterial) or use the name purely for marketing purposes (Jones et al., 2015b). Figure 2 shows a product that is marketed as a ‘nano-coating’ on wood surfaces, but does not actually contain nanoparticles.

Where products do contain nanomaterials and state this, many do not give information regarding the specific nanomaterials used or the quantity contained (Jones et al., 2015a). Safety data sheets in the UK and Europe are not required to specify the presence of nanomaterials, as they need only list substances that would be categorised as hazardous under the Classification, Labelling and Packaging (CLP, 2009) regulations. Figure 3 shows an extract from a safety data sheet for a coating product; CNTs are not listed in the ingredients, even though the product is marketed as a coating that “…combines CNT in an epoxy-polyamide formulation”. Several authors have reported that safety data sheets rarely, if ever, contain details regarding nanomaterials (Larraza et al., 2015), and in fact they may actually contain incorrect data. For example, Lee et al. (2012) reported finding safety data sheets for CNTs that presented the published hazard data for graphite – a material that is chemically the same but very different in terms of structure and toxicity.

2.5 ‘Nanoparticle exposure is predictable’

Predicting nanoparticle release and the extent to which workers might be exposed is difficult, beyond the basic

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Figure 2. SEM image of a coating marketed as a ‘nano-coating’ but contains no nanoparticles.
principle that particle release from stable solid materials is negligible and that release from aerosols or powders is potentially much higher (Baron, 2015; Hansen et al., 2014). There are several reasons for this.

- There is a lack of certainty regarding which products might contain nanomaterials, as discussed earlier. In addition, nanomaterials may be released in a form different from that in which they were added – particles can be changed through chemical reactions or through agglomeration or aggregation.

- Weathering or machining of products may increase the potential for particle release, but there is some evidence (although incomplete and limited) that the nanomaterials released are often still bound to the underlying matrix rather than being free nanoscale particles (Froggett et al., 2014; West et al., 2016).

- Nanoparticles can be released from products that did not intentionally contain nanomaterials – thus the demolition of traditional concrete produces particles of many sizes, including the nanoscale (Azarmi et al., 2014, 2015).

In practice, therefore, the only way to assess nanoparticle exposure is to measure it, quantifying particle release from commercially available products in real-world scenarios.

### 2.6 Released nanoparticles are easy to measure

Accurate assessment of worker exposure (e.g. to dust or aerosols) is important, as it influences the ‘dose’ an individual is likely to receive, which in turn predicts toxicity. The principle that ‘the dose makes the poison’, attributed to Paracelsus writing in 1564 (Deichmann et al., 1986) is as true for nanomaterials as it is for other substances. Traditionally, dust or particle release is measured using mass-based methods, and recommended exposure limits are given in mg/m³. Although these techniques can be used for nanomaterials, they are relatively insensitive (Donaldson and Poland, 2013; Kuhlbusch et al., 2011; Savolainen et al., 2010). Figure 4 shows how the presence of nanomaterials can have a relatively small impact on the mass of a dust sample, and thus may be missed or considered insignificant when using mass-based approaches. An alternative method is to count the number of particles, which is much more sensitive. However, the equipment required to do this is relatively new; it is generally too large to be easily portable and is particularly unsuited to taking measurements of individual workers’ exposure (Methner et al., 2009a). In addition, it can be difficult to differentiate between the nanoparticles released by a particular process or activity and those in the background, for example, from things such as tobacco smoke (van Broekhuizen et al., 2011) or exhaust fumes (Kumar and Morawska, 2014).

### Table 3: Composition/information on ingredients

<table>
<thead>
<tr>
<th>Chemical name</th>
<th>CAS#</th>
<th>EINECS#</th>
<th>EU classification (67/548/EEC)</th>
<th>CLP annex VI classification</th>
<th>%</th>
</tr>
</thead>
<tbody>
<tr>
<td>Epoxy resin</td>
<td>25068-38-6</td>
<td>500-033-5</td>
<td>Xi, N R36/38, R43, R51/53</td>
<td>Skin Irrit. Cat 2 (H315), Eye Irrit. Cat 2 (H319), Skin Sens. Cat 1 (H317), Ac. Chronic Cat 2 (H411)</td>
<td>35–45</td>
</tr>
<tr>
<td>Talc</td>
<td>14807-96-6</td>
<td>238-877-9</td>
<td>No EU classification or R phrase</td>
<td>No GHS Classification</td>
<td>25–35</td>
</tr>
<tr>
<td>1,2,4-Trimethylbenzene</td>
<td>95-63-6</td>
<td>202-436-9</td>
<td>Xi, Xn, N R10, R20, R36/37/38, R51/53</td>
<td>Flam. Liq. Cat 3 (H226), Acute Tox. Cat 4 (H352), Skin Irrit. Cat 2 (H315), Eye Irrit. Cat 2 (H319), STOT SE Cat 3 (H335), Ac. Chronic Cat 2 (H411)</td>
<td>&lt;5</td>
</tr>
</tbody>
</table>

Figure 3. Extract from a safety data sheet for a coating marketed as containing CNTs. The CNTs are not mentioned anywhere in the safety data sheet.
The current best practice is to measure nanoparticle exposure using a combination of mass- and particle-based methods. This should be supplemented by the use of scanning electron microscopy (SEM) or transmission electron microscopy to enable characterisation of any particles (e.g. what substance they are, whether they are bound together, what shape they are) (Bard et al., 2015; Methner et al., 2009b; Vaquero et al., 2015).

3. Common nano-enabled construction materials

A database of 156 nano-enabled construction products has been developed through a combination of academic and manufacturers’ literature, interviews with stakeholders in construction, demolition and nanomaterial production (n = 55) and laboratory testing (Table 1).

The following section provides more details about these products. As discussed previously, it is difficult to identify with certainty products that are nano-enabled and/or the nanomaterials contained. Consequently, the criteria for including products in Table 1 are broad, and include claims made by manufacturers regarding nanomaterial use – for example, the term ‘nano’ in a product title or literature, information from academic reports and assumptions based on product properties.

3.1 Coatings

Nano-enabled coatings are widely available, both for the home improvement market and for professionals. In addition, products such as floors, toilets and roofs can be purchased that have been pre-treated with nano-based coatings. Coatings are the dominant and most numerous nano-enabled products (Lippy and West, 2015; van Broekhuizen et al., 2011; West et al., 2016), accounting for over half the current and predicted spending on construction nanomaterials (Freedonia, 2007). However, actual use relative to non-nano products remains low, with estimates suggesting they account for no more than 1% of the coatings market in Europe (Hincapie et al., 2015). The nanomaterials commonly used in coatings include silica, titanium dioxide, silver and carbon.

Nanosilica is used in various forms to provide water repellancy, abrasion resistance and ultraviolet protection (Pagliaro et al.,...
2009). Greßler et al. (2010) describe the lotus leaf effect, where a surface that appears smooth is in fact covered in tiny protrusions that prevent water drops from adhering to the surface (Greßler et al., 2010). Other nanosilica-based coatings have different modes of action. Figure 5 illustrates the high contact angle of water drops on a concrete surface that has been coated and is now superhydrophobic.

Titanium dioxide is used for its photocatalytic effects. In the presence of ultraviolet light, dirt and particles as well as pollutants such as nitrogen oxides are broken down into water, carbon dioxide and nitrates. At the same time, the hydrophilic effect of the coating enables water to spread over the surface and wash away any waste products (Greßler et al., 2010). Thus, surfaces become ‘self-cleaning’, and there is also potential for the reduction of airborne pollution (Chen and Poon, 2009; Shen et al., 2012).

Silver is added to coatings for its anti-microbial effects, and marketed particularly for use in health care and food preparation environments. It is the nanomaterial most often listed as the active constituent in commercial products (PEN, 2014) as well as being the nanomaterial most commonly reported or recognised by those working in construction (Arora et al., 2014; Hincapié et al., 2015).

Carbon nanomaterials can be used in coatings. One CNT-based coating has been identified that is marketed for its ability to reduce corrosion. It is expensive and likely to be used only in specialist applications. CNTs might also be used for their fire resistance and their ability to reduce fouling in marine environments, but it is difficult to find evidence of commercially available products of this nature. Other carbon-based nanomaterials are used more widely, such as carbon black, sold as a shield against radio waves. Graphene is reportedly also used, although, as the SEM of a coating product in Figure 6 illustrates, the particles are relatively large, around 40 nm thick, and several microns in the other two planes. They thus fall within the formal definition of ultra-thin graphite (Sanchez et al., 2011) rather than true graphene, which would typically be no more than 3 or 4 nm thick.

The lack of relevant information in safety data sheets makes it extremely difficult to identify the exact nanomaterial used in individual products. Table 2 provides some examples of nano-enabled coatings using information that is available publicly, although not always easy to find. Neither is information freely available on the proportions of nanomaterial used. For example, a review by Munafò et al. (2015) of titanium-based coatings used in stone conservation shows concentrations ranging from 0·3 to 10% (by weight) and also shows that products vary in their effectiveness in terms of self-cleaning, biocidal effects, anti-pollution effects and longevity. Other studies have shown variation in the effectiveness of coatings in preserving wood (Marzi, 2015) and the bactericidal effect of nanosilver (Molling et al., 2014). This lack of consistency is not limited to construction – nanosilver is added to socks to provide anti-odor and anti-microbial properties, but the quantities used have been shown to vary 700 fold between different brands (Benn and Westerhoff, 2008).

3.2 Glass

Nanomaterials can be applied to sheets of glass as part of the manufacturing process, and then used to construct window units with special properties. Multiple layers might be added to

![Figure 5. Water drops on a nano-coated concrete brick, showing how the drops form a near-spherical shape.](image)

![Figure 6. SEM image of a product marketed as ‘graphene’, but more accurately described as ultra-thin graphite or nanographite.](image)
provide various properties within one unit, but the total coating layer remains very small, around 15 nm thick according to the manufacturer’s literature (Pilkington, 2016).

Nanomaterials can provide windows with self-cleaning and thermal insulation properties (Table 3). They are also used to reduce solar gain in hot environments and can be used in switchable windows, where the amount of heat and light passing through a window can be controlled by the user.

3.3 Concrete

Concrete is widely discussed in academic literature as a nano-enabled material, with many papers describing the use of CNTs in a laboratory environment to improve the materials’ strength, or to provide electrical conductivity (summarised, e.g. by Raki et al., 2010; Sanchez and Sobolev, 2010; Siddique and Mehta, 2014). CNTs can be added to concrete, but their commercial application is less advanced than might first appear from the literature (Table 4). CNTs are expensive, particularly when considering the quantities required for addition to mass concrete, even though typical recommended proportions are only 1 wt%. They are also very difficult to disperse uniformly in concrete (Raki et al., 2010). Very few products appear close to commerciality, although one example has recently been approved for use by the Georgia Department of Transport in the USA (EE, 2015b).

More widely used in concrete is silica. Silica fume is an amorphous (non-crystalline) polymorph of silicon dioxide (silica). It is an ultrafine powder (with a typical average particle size of 150 nm) collected as a by-product of silicon and ferro-silicon alloy production (Goodier et al., 2016, unpublished). It has been available as a nanoscale concrete additive for over 30 years (ACI, 1996) and might be used where enhanced durability, surface finish or early high strength is required. Nanosilica is an additional concrete additive, with a particle size around one tenth of that of silica fume (Sanchez and

<table>
<thead>
<tr>
<th>Nanomaterial</th>
<th>Properties attributable to nanomaterial use</th>
<th>Availability and extent of application</th>
</tr>
</thead>
<tbody>
<tr>
<td>Silica (silicon dioxide) (including silane, siloxane and polysilazane; some products refer to the presence of nanoparticles, others state that they are not particulate)</td>
<td>Water repellancy, Easy cleaning, stay clean, Graffiti resistance, Hardening, protection, Scratch resistance</td>
<td>Many products are available with these properties. Some are explicit about being nano-enabled, although they may not be specific about the substance used; these include ‘Nanoclean cleaner’ (Adseal), ‘ThermoSan’ (Caparol), ‘Surfapore’ (Nanophos), ‘Nansulate’ (Industrial Nanotech) and ‘Nanostone and Nanowood’ (Nanoprotect)</td>
</tr>
<tr>
<td>Titanium dioxide</td>
<td>Protective, anti-bacterial, Anti-pollution, Self-cleaning</td>
<td>Many products are available with these properties and are explicit about their nano-enablement – for example, ‘FN coating’ (FN NanoInc), ‘KNOxOut’ (Boysen), ‘P and T-230’ (nanoprotect) and ‘Surfatix’ (nanophos)</td>
</tr>
<tr>
<td>Silver</td>
<td>Anti-microbial</td>
<td>Many products are available with these properties and are explicit about their nano-enablement – for example, ‘NPS 100, 200’ (Nanophos); ‘Bioni roof, Bioni hygienic, Bioni nature’</td>
</tr>
<tr>
<td>CNT</td>
<td>Resistance, anti-corrosion</td>
<td>‘Teslan’ (Tesla nano) is marketed as a coating that contains CNTs. Other CNT products are marketed to be added to coatings but are not necessarily available in commercially available coatings</td>
</tr>
<tr>
<td>Other carbon</td>
<td>Electromagnetic shielding, Anti-condensation, Heating paint</td>
<td>Products likely to be based on carbon black are relatively cheap and have been available for many years. They include ‘Carbo e-therm’, ‘Carbosil’ (Future carbon) and ‘Bloc paint’ (Craig and Rose) ‘Graphenstone’ is marketed as a product that contains graphene</td>
</tr>
</tbody>
</table>

Table 2. Nanomaterials in coatings
Sobolev, 2010), used to produce very high strength concretes. It is expensive and rarely used in commercial concreting operations.

Nanotitanium, described earlier for its photocatalytic effect in coatings, can also be used as a concrete additive, either throughout the concrete itself or applied as a separate coating layer (as it is only effective at the surface of the material where it can be exposed to light). White surfaces remain white as a result of this self-cleaning effect (Greßler and Gazsó, 2012), and there is some evidence of improved air quality through its use on pavements, tunnels and walls, provided suitable light is available (Allen et al., 2008; Guerrini, 2012; Shen et al., 2012). Although these products have been commercially available for many years (Zhu et al., 2004), there is limited evidence of their use in practice outside of showcase building projects and research studies.

### 3.4 Steel

The enhancement of steel comes from the use of nanotechnology rather than the addition of nanoparticles; processes are used that drive out impurities and refine grain boundaries to reduce corrosion or increase strength (Kolpakov et al., 2007; Shi, 2010). Changes can be made to the steel itself, or can be used for coatings that work in a similar way (Shi, 2010) (Table 5). Composites of nano-enabled steel and ceramics can be used to produce materials with high wear resistance and strength (Rodriguez-Suarez et al., 2012).

#### 3.5 (Aerogel) insulation

Silica aerogels are a highly effective insulator (AA, 2014; Cuce et al., 2014) consisting of 97% air in a silica framework. Their nanoporous structure (manufacturers state that they do not contain nanoparticles) provides resistance to heat flow (Hanus and Harris, 2013). Aerogel-based products exist in three main forms (Table 6). They are much thinner and lighter than traditional materials for an equivalent level of thermal insulation and are therefore useful where there is limited space, for example, in retrofit and refurbishment projects. They are also used in challenging industries such as oil and gas. However, the products are expensive, typically 6–10 times the cost of more traditional insulation products (Cuce et al., 2014), and are therefore not frequently used in standard building projects.

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**Table 3. Nanomaterials in window glass**

<table>
<thead>
<tr>
<th>Nanomaterial</th>
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</thead>
<tbody>
<tr>
<td>Titanium dioxide</td>
<td>Self-cleaning</td>
<td>Windows with these properties are readily available from the main glass suppliers – for example, ‘Activ’ (Pilkington) and ‘Biodine’ (SGG). These are often used on conservatories and glass roofs.</td>
</tr>
<tr>
<td>Metal oxide coating</td>
<td>Thermal insulation (low emissivity; low-e glass)</td>
<td>Windows offering these properties are readily available from the main glass suppliers – for example, ‘Optitherm’ (Pilkington), ‘Planitherm’ (SGG) and ‘Clima guard’ (Guardian). They are used to ensure good insulation and aid compliance with building regulations. Nanomaterials on windows are referred to as ‘soft’ coatings; some window manufacturers favour older, non-nano versions (known as ‘hard’ coatings) as these are more durable (Jelle et al., 2012)</td>
</tr>
<tr>
<td>Metal oxide coating</td>
<td>Solar control</td>
<td>Windows with these properties are readily available from the main glass suppliers – for example, ‘Suncool’ (Pilkington), ‘Cool-lite’ (SGG) and ‘Sunguard’ (Guardian). These are used to reduce solar gain. They can be used in conjunction with low-e coatings for internal climate control</td>
</tr>
<tr>
<td>Silicon dioxide</td>
<td>Fire safety</td>
<td>Products that contain an intumescent layer of precipitated nanosilica between two panes of glass have been available for up to 30 years for use in high-risk environments such as fire escape routes. Likely examples of this include ‘Pyrostop, Pyrodur’ (Pilkington), ‘Pyranova’ (Schott) and ‘Pyrogard’ (CGI)</td>
</tr>
<tr>
<td>Tungsten dioxide</td>
<td>Smart switchable</td>
<td>These allow the amount of light and heat passing through a window to be varied by the user. They are relatively novel and expensive, and not widely used outside of specialist projects. Examples of these include ‘E-Control glas, View glas and Sage glass’</td>
</tr>
</tbody>
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3.6 Other products

Nanotechnology and nanomaterials are important in the development of photovoltaics, using smaller quantities of materials than traditional silica wafers. Some are available commercially, which use different forms of silica; others use nanoscale layers of materials such as cadmium, telluride, selenium and copper (Jelle et al., 2012; Khan and Arsalan, 2016; Pagliaro et al., 2008). In addition, solar cells based on other nanomaterials such as titanium, perovskite, quantum dots and graphene are at various stages of development and commercialisation.

Buildings may also contain other products that use nanotechnology, including radiator fluids based on metal nanoparticles (Alawi et al., 2014) (e.g. hydromx) and electronic devices such as batteries, organic light emitting diodes (oleds) and computer screens.

Nanoclays, which are either naturally occurring or synthetically created, consist of layers of silicate-based materials (Morgan, 2007). They can be used in polymers to improve functionality in various ways, but there are no apparent construction products available commercially.

4. The future of nanomaterials in construction – the risks and benefits

This paper has presented an overview of the nanomaterials currently commercially available for use in construction. There are clearly benefits associated with the introduction of many nano-enabled products, but there may also be hazards, as health risks are associated with some nanomaterials. The exact form and the dose involved are important determinants of whether the risk is significant, making it important that there is a good understanding of the materials being used. However, reviewing the limited information available in the public domain has shown that it is difficult to be certain regarding the extent of nano-enablement in construction products and their market penetration. It is even more challenging to predict which new developments in nanomaterials may
become commercially available for products in the near (or far) future and how widely they might be used. Current evidence regarding the risk potential from nano-enabled construction products suggests that current construction worker exposure to nanomaterials is likely to be relatively low (Larraza et al., 2015) the presence of existing hazards in coatings (such as solvents) is generally more significant than any specific risks from nanomaterials (Vaquero et al., 2015) the measures recommended to protect against existing risks in construction (e.g. water suppression, dust extraction at source, use of (face-fitted) FFP3 masks where appropriate (HSE, 2013a)) will in most cases be effective against the additional risks from nanomaterials (Boutry et al., 2015; Methner et al., 2009a).

However, it is important to be alert to the increased use of products that has been predicted (e.g. AECOM, 2014; Freedonia, 2007). This could involve: greater use of products that are already available, such as concretes based on amorphous silica or windows with self-cleaning or thermal coatings; an increased availability of products using potentially hazardous materials such as CNTs; or the introduction of products that use new materials with unknown toxicity profiles. The possible health implications of these scenarios are clearly diverse.

### Table 5. Nanomaterials in steel

<table>
<thead>
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<tbody>
<tr>
<td>Nanostructured steel or steel alloys</td>
<td>High strength, corrosion resistance</td>
<td>A number of products are available, and have been marketed for at least 10–15 years. These include ‘ChromX’ (Mmfx), which is reportedly used in rebars for road construction and in other civil projects (Larsen, 2013) Other products such as ‘AHSS’ and ‘NSB’ (Nanosteel) are marketed more for the automotive industry and for high-demand environments such as mining and oil and gas</td>
</tr>
<tr>
<td>Ceramic metal composite</td>
<td>Hardness, repair of damaged metal surfaces, high wear resistance, corrosion resistance</td>
<td>Products such as ‘Cermet’ and ‘PCompP’ (Mesocoat) may be used in engines or motors used for construction, or in high-risk industries such as oil, gas and so on</td>
</tr>
<tr>
<td>Metallic coatings</td>
<td>High strength and toughness in welding, high wear resistance</td>
<td>Examples include ‘SHS’ (Nanosteel) and ‘Santronic’ (Sandvik). These are marketed, for example, for the coating of concrete chutes to prolong their life</td>
</tr>
</tbody>
</table>

### Table 6. Nanomaterials in aerogel insulation materials

<table>
<thead>
<tr>
<th>Nanomaterial</th>
<th>Properties attributable to nanomaterial use</th>
<th>Availability and extent of application</th>
</tr>
</thead>
<tbody>
<tr>
<td>Silica aerogel in translucent particles</td>
<td>‘Lumira’ particles (Cabot) are available on their own or incorporated into wall/ceiling products such as ‘Ecosky, Kalwall, Solera’. They allow light transmission while still providing a high level of thermal insulation</td>
<td>These products are readily available on the market but their use is largely confined to specialist projects as they are much more expensive than traditional products</td>
</tr>
<tr>
<td>Silica aerogel in blanket form</td>
<td>These are available from several manufacturers: some are specifically designed for protection against cold, others have high heat resistance; products include ‘Pyrogel, cryogel, space loft’ (Aspen aerogels), ‘Thermal wrap’ (Cabot) and ‘Thermablok’ (Acoustiblok)</td>
<td></td>
</tr>
<tr>
<td>Silica aerogel particles in vacuum insulated panels</td>
<td>Vacuum insulated panels are available for installation in walls, flooring and roofs, products include ‘Optim-R’ (Kingspan) and ‘Kevothermal’</td>
<td></td>
</tr>
</tbody>
</table>
Several surveys in Europe and the USA (Mann, 2006; Teizer et al., 2009; van Broekhuizen and van Broekhuizen, 2009; West et al., 2016) have shown that construction professionals often have limited knowledge of this field, which is not surprising given the many sources of uncertainty and lack of clarity discussed previously. It is nevertheless important that construction professionals ask searching questions of their suppliers when they are using new products. Where there is insufficient information available to judge the hazardousness of a material or product, it is recommended that a ‘precautionary approach to risk management’ is adopted (HSE, 2013b). This involves erring on the side of caution when assessing risk, particularly considering the possible consequences of an event even if the likelihood is low. For example, if a product contains CNTs but there is insufficient evidence to determine their shape, size and toxicity, it might be reasonable to assess risk as if they are a hazardous form, unless evidence is available to the contrary.

A high awareness of the new products being developed will also enable the construction industry to take full advantage of the benefits that nanotechnology can offer. With better understanding, there might be an increase in the use of products that are already available but rarely specified – for example, insulation that is four times as effective as traditional materials, photocatalytic concretes that could reduce airborne pollution or high-performance concrete that reduces the need for steel reinforcement. It might also encourage the early adoption of products that are on the horizon but not yet fully developed – for example, vanadium-based coatings that eliminate the need for window cleaning on high-risk buildings, or photovoltaic fabrics.

Nanomaterials offer both benefits and risks. They are present in some commercially available construction products now, some of which are labelled and some of which are not, and there are undoubtedly more to come. It is important that the construction industry knows enough about them to make informed, smart and safe decisions.

REFERENCES


HSE (Health and Safety Executive) (2013a) Construction Dust Construction Information Sheet No 36 (Revision 2). Health and Safety Executive, Sudbury, UK.

HSE (2013b) Using Nanomaterials at Work. Health and Safety Executive, Sudbury, UK.


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