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Combinatorial study of Sn-Ti-W-O transparent conducting oxide thin films for photovoltaic applications

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Abstract — a combinatorial study of transparent conducting oxide thin films based on SnO₂–TiO₂–WO₃ phase space is reported. These multinary oxide films were fabricated by magnetron reactive co-sputtering of tin monoxide (SnO), titanium (Ti) and tungsten (W) targets. SnO₂–TiO₂–WO₃ film compositions with Ti/Sn ratio (0.02 – 0.12) and W/(Ti+Sn) ratio (0.02 – 0.25) were explored. The effect of oxygen partial pressure on composition, structure and optical properties was evaluated. High optical transparency above 80% across the visible spectrum was obtained for sputtered ternary SnO₂–TiO₂ oxide films for oxygen partial pressure > 19.4%. A positive correlation between optical bandgap and Ti/Sn ratio was observed. However, optical properties deteriorated as Ti-content increased in the as-deposited SnO₂–TiO₂–WO₃ films. All studied as-deposited SnO₂–TiO₂–WO₃ thin films were found to be highly resistive. X-ray diffraction data indicated no long-range structural order.

Index Terms — transparent conducting oxides, amorphous thin films, magnetron reactive sputtering, combinatorial technique

I. INTRODUCTION

Transparent conducting oxides (TCOs) perform critical functions in a wide range of opto-electronic devices. For instance, thin-film solar cells have traditionally used doped wide-bandgap TCOs such as SnO₂:F, In₂O₃:Sn and ZnO:Al as transparent conducting electrodes, anti-reflection coatings and chemical barriers, etc. [1]–[6]. However, these crystalline TCOs need elevated deposition temperatures to optimize their physical, optical and charge-carrier transfer properties. This is a severe handicap for temperature-sensitive fabrication of next-generation organic and flexible photovoltaic devices. Amorphous TCOs thin films have been reported as potential alternatives to their conventional crystalline counterparts [7]–[9]. The approach follows the ability to tailor the physical, optical and charge-carrier transfer properties over a range of atomic compositions in the candidate amorphous thin films [10]–[13]. More recent studies of multicomponent oxides on InZnO [14], ZnO-In₂O₃-SnO₂ [15] and ZnSnO [16] have enunciated this concept. Potential amorphous TCO candidates include the nontoxic and environmental-friendly TiO₂–SnO₂–WO₃ thin films. TiO₂ and SnO₂ have similar crystal structures, chemical bonds, bond-lengths and band energies [17]–[19]. Both are also remarkably stable oxides. Earlier reports on magnetron sputtered WO₃ thin films have shown the possibility of modulating the physical, optical and electronic by judicial control of depositions conditions [20]–[22]. Ti-doped WO₃ [23], [24] and W-doped TiO₂ [25] studies have shown considerable possibilities that W–Ti oxide films offer in optimizing the TCO thin-film structure, physical and opto-electronic properties for high-efficiency and low-cost next-generation photovoltaic devices. Magnetron reactive co-sputtering technique [26] provides opportunity of producing TiO₂–SnO₂–WO₃ thin films with controllable stoichiometry and composition. This report presented here investigates the structural and optical properties of sputtered TiO₂–SnO₂–WO₃ thin films in relation to the composition and the effect of reactive gas (O₂) pressure.

II. EXPERIMENTAL

Film synthesis - compositionally graded Sn-Ti-W-O thin films were deposited by magnetron co-sputtering of ceramic SnO-target (99.99% purity), Ti-target (99.99% purity) and W-target (99.99% purity). The films were deposited using an AJA International Orion 8HV sputter coater onto standard RCA pre-cleaned soda-lime glass (SLG) substrates (size 10cm × 10cm) at room temperature. A metallic mask was used to create a 5×5 matrix with each sample size of 1.8cm × 1.8cm. The sputtering chamber base pressure was maintained below 10⁻⁷ Torr. The working pressure and Ar (sputtering gas) flow rate of 1mTorr and 5SCCM respectively, were selected after preliminary investigations to determine the optimum deposition conditions. An oxygen (O₂ 100% purity) supply provided the reactive gas. The O₂ flow rate was varied to control the oxygen partial pressure during depositions. Independent RF, DC and pulsed DC power sources were used to excite a three-gun assembly, shown in Fig. 1.
The target power was adjusted to control the sputtered flux condensing rate on stationary substrates. The coating duration was set to give the desired film thickness. Prior to each coating, each target was presputtered for 120s (with shutter closed) to remove target surface contaminants. All Sn-Ti-W-O thin films were deposited at room temperature and analyzed as-deposited.

**Film analysis** - At each coordinate on the 5 × 5 matrix, the analysis focused on structure, composition and optical properties. This allowed for rapid correlation of the results. Crystal structure was measured using a Bruker D2 phaser X-ray diffractometer with a Cu-Kα X-ray source and Lynxeye™ detector. A Cary 5000 UV-VIS-NIR spectrophotometer was used to determine the optical properties. An AMBIOS XP-2 profilometer was used to determine the film thickness. ThermoFisher Scientific™ K-Alpha™ X-ray photoelectron spectrometer (XPS) was used to probe the elemental contents of the samples.

**III. RESULTS AND DISCUSSION**

**A. RT magnetron sputtered SnO₂ and TiO₂ thin films**

**Optical properties** - The dependence of the deposition rate, physical and optical properties of RT sputtered SnO₂ thin films on oxygen partial pressure (ppO₂) was investigated. Fig. 1 shows behavior of the deposition rate, optical transmittance (in the visible spectrum) and optical bandgap with sputtering reactive gas.

The SnO target power was set at 180W (RF supply). The sputtered SnO₂ thin films exhibited an average thickness of 440nm. The average deposition rate decreased from a maximum of 12.5nm/min at ppO₂ of 0% (100% Ar plasma) to less than 3nm/min for ppO₂ > 40%. The deposition rate rapidly decreases when even small amounts of O₂ are added to an Argon-plasma. This decreasing behavior has previously been reported when a ZnO and SiO₂ targets have been sputtered using various Ar/O₂ gas mixtures [27][28]. However, optical properties of these films improved with the increase in ppO₂. Optical transmittance in the visible spectrum of SnO₂ films increased from 53% at ppO₂ of 0% to 85% for a ppO₂ of 40%. This trend is mirrored by the optical bandgap behavior.

TiO₂ thin films were deposited using a metallic Ti target (180W Pulsed DC). Fig. 2 shows variation of TiO₂ film deposition rate, average VR transmittance and optical bandgap with O₂ partial pressure (ppO₂).

There is a markedly abrupt drop in the TiO₂ deposition rate at the ppO₂ of 40%. The deposition rate at averages 7.5nm/min for ppO₂ < 40% and then substantially reduces to below 2nm/mm thereafter. This sudden reduction in growth rate is most likely a result of TiO₂ formation on the Ti target surface [29]. The Ti target transitions from operating in “metallic mode” with high yield rate to “oxide mode” low yield mode. The TiO₂ thin films deposited in “metallic mode” conditions were less optically transparent (average VR transmittance below 10%) compared to films deposited in the “oxide mode” region with average VR transmittance of 80%.

**Fig. 2. Room temperature reactive sputtering of SnO₂ thin films**

**Fig. 3. Room temperature reactive sputtering of TiO₂ thin films**

**Fig. 4. XRD patterns of RT as-deposited SnO₂ and TiO₂ thin films**
Film structure – X-ray patterns of as-deposited SnO$_2$ and TiO$_2$ films deposited at oxygen partial pressure (ppO$_2$) levels of 7.4%, 19.4% and 32.4% are shown in Fig. 4. Apart from the “broad-hump” shape from the glass substrate shown between 20° and 40°, no sharp diffraction peaks can be seen in all the samples. These profiles are indicative of the absence of long-range structural order in the RT sputtered films. However, it is difficult to determine whether the as-deposited SnO$_2$ and TiO$_2$ films are amorphous or composed of nano-crystallites of only few nanometers which cannot be detected by XRD, because both structural phases exhibit a broad X-ray diffraction peak.

B. RT magnetron sputtered SnO$_2$-TiO$_2$ thin films

Fig. 5 shows the colour map of Ti/Sn ratio of a 5 x 5 spatial matrix for room temperature co-sputtered Ti (60W DC power) and SnO (60W RF power). The deposition conditions were set at pressure of 1 mTorr, Ar-flow of 5SCCM and oxygen partial pressure (ppO$_2$) of 24.2%. The average thickness of the sample was 270nm. XPS analysis showed Ti atomic percent with a maximum at 12.5% and high Sn atomic percent varying between 87.4% and 96.9%.

The Ti/Sn atomic ratio varies from 0.04 to 0.125 across the Sn-Ti-O film sputtered at a ppO$_2$ of 24.2%. The average Ti/Sn atomic ratio corresponds to the deposition rates of sputtered TiO$_2$ and SnO$_2$ fluxes on the substrate surface.

Chemical composition and optical bandgap

Fig. 6 (a) and Fig. 6 (b) Ti/Sn atomic ratio and optical bandgaps at selected coordinates (1,1), (1,2), (1,3) and (1,4) and (2,1), (2,4) and (2,5) respectively. The graphs show a positive correlation between the Ti/Sn atomic ratio and optical bandgap in Sn-rich compositions. This result is in good agreement with optical transmittance data obtained from spectrophotometry and the literature [19], and provides a quick way of determining the composition of the SnO$_2$-TiO$_2$ films.

C. RT magnetron sputtered SnO$_2$-TiO$_2$-WO$_3$ thin films

Chemical composition and optical properties

X-ray photoelectron spectroscopy was used to obtain the chemical composition and valence state of the elements present in the grown films. The XPS survey spectra of the films identified the main constituents as Sn, Ti, W, O, and C. The carbon peak seen in the XPS spectra was due to carbon from air exposure before being placed in the XPS system.

Fig. 8, Fig.9 and Fig.10 show composition colour maps of the sputtered SnO$_2$-TiO$_2$-WO$_3$ thin films deposited by co-sputtering of W-target (120W RF), Ti-target (105W Pulsed DC) and SnO-target (60W DC). The sputtering pressure was set at 1mTorr, with Ar-flow of 5SCCM with oxygen partial pressure (ppO$_2$) of 24.2%. The average thickness of the deposited film was 492nm. XPS analysis showed average Ti
atomic percent of 0.46%, Sn atomic percent of 8.87% and W atomic percent of 11.54%.

Fig. 8. Compositional mapping of the ratio W/(Sn+Ti) of a 10cm × 10cm; each [x,y] corresponds to one 1.8cm × 1.8cm cell in SnO₂-TiO₂-WO₃ thin films

Fig. 9. Compositional mapping of the ratio Ti/(Sn+W) of a 10cm × 10cm; each [x,y] corresponds to one 1.8cm × 1.8cm cell in SnO₂-TiO₂-WO₃ thin films

Fig. 10. Compositional mapping of the ratio Sn/(Ti+W) of a 10cm × 10cm; each [x,y] corresponds to one 1.8cm × 1.8cm cell in SnO₂-TiO₂-WO₃ thin films

Fig. 11 and Fig. 12 show optical transmittance and reflectance curves for respective coordinates on the matrix grid of the SnO₂-TiO₂-WO₃ thin films. These films exhibited high transmittance (>80%) across the visible spectrum. No free carrier absorption effects were observed. This was reflected in high sheet resistances observed in as-deposited SnO₂-TiO₂-WO₃ films. Fig. 12 shows a marked decrease in transmittance as Ti atomic content increased.

Fig. 11. Optical transmittance and reflectance plots of as-deposited SnO₂-TiO₂-WO₃ thin films (at coordinates (1,1), (1,2), (1,4) and (1,5))
IV. SUMMARY

Multinary oxide SnO$_2$-TiO$_2$-WO$_3$ thin films were fabricated by reactive co-sputtering of SnO, Ti and W targets. The structural and optical properties of compositional ranges Ti/Sn of (0.02 – 0.12) and W/(Ti+Sn) of (0.02 - 0.25) were explored. XRD data of the as-deposited SnO$_2$-TiO$_2$-WO$_3$ thin films reveal the absence of long-range structural order. It was observed that film deposition rate for Ti, W and SnO targets declined with an increase in O$_2$ partial pressure. Sputtered films Optical transmittance in the visible spectrum of SnO$_2$ films increased with oxygen partial pressure and this correlated well with the optical bandgap behavior. Optical properties deteriorated as Ti-content increased in the as-deposited SnO$_2$-TiO$_2$-WO$_3$ films. All studied as-deposited SnO$_2$-TiO$_2$-WO$_3$ thin films were found to be highly resistive.

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