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Data Article

Experimental data in support of characterization of the CePO₄ dispersion into transparent PMMA/PU IPNs by the sequential route


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Abstract

This article is focused on the complementary data referring to the article “Dispersion of upconverting nanostructures of CePO₄ using rod and semi-spherical morphologies into transparent PMMA/PU IPNs by the sequential route”. It contains the XPS data of CePO₄, photographs and DSC thermograms of transparent PMMA/PU IPNs as well as with CePO₄ dispersed in different wt.%, Confocal laser scanning micrographs, transmission electron microscopy (TEM), optical images of surface, and visual inspection (photographs) before and after aging of hybrid materials.

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### Value of the data

- This data can be useful for comparing the dispersion of CePO₄ nanostructures in similar systems and how the morphology affects the structural, thermal and mechanical properties.
- Data highlights the influence of dispersed CePO₄ in different PMMA/PU ratios.
- This article will serve as a guideline to select the amounts adequate for dispersing CePO₄ into PMMA/PU.

### 1. Data

Plastic materials with good durability can be obtained from the synthesis of interpenetrating polymer networks (IPNs) while incorporating luminescent nanostructures such as Fluoro-functionalized nanostructured silica (FSiO₂), silica (SiO₂), carbon black (CB), barium titanate (BaTiO₃), polyaniline, and cerium phosphate (CePO₄) [1–6].

The dataset of this article shows additional information about the dispersion of CePO₄ in rods and semi-spherical morphologies, into transparent PMMA/IPN. Initially, this study provides facts of the elements in CePO₄. For this reason, Fig. 1 shows the X-ray spectroscopy (XPS) spectra of nanorods and semi-spherical particles of CePO₄. This data also include photographs of different ratios of PMMA/PU that showed the transparency of the samples synthesized (Fig. 2). The thermal properties of PMMA/PU IPNs are confirmed in thermograms of selected PMMA/PU IPNs in Fig. 3. In addition, visual examination after incorporation of nanostructure in the polymer was analyzed. The incorporation, 0.1, 0.5, and 1 wt.% of CePO₄ (nanorods and semi-spherical) in 50/50 ratio of PMMA/PU IPNs is shown in photographs (Fig. 5). In this context, dispersion and emission properties of CePO₄/PMMA/PU IPNs is confirmed by confocal laser scanning images showed in Figs. 6 and 7. Dispersion of nanorods and semi-spherical CePO₄ was evaluated in a selected sample, 50/50 PMMA/PU IPNs (Fig. 8). On the other hand, the texture of fractures from the tensile test is observed in Fig. 9. Finally, visual inspection before and after accelerated weathering test is seen in Fig. 10.
2. Experimental design, materials, and methods

2.1. Cerium phosphate (CePO₄) synthesis and characterization

CePO₄ nanostructures in nanorod and semi-spherical morphologies were obtained at pH 1 and 11, respectively, by the microwave-assisted hydrothermal method following the procedure described in Ref. [7].

Fig. 1. XPS spectra of nanorods and semi-spherical CePO₄ nanoparticles.

Fig. 2. Photographs of pure PMMA/PU IPNs in different ratios.
Fig. 3. DSC thermograms of PMMA/PU IPNs.

Fig. 4. Scheme of polymerization process of hybrid PMMA/PU/CePO4 IPNs.

Fig. 5. Photographs of pure PMMA/PU IPNs (50/50 system) and PMMA/PU/CePO4 IPNs (50/50 system) with the addition of nanorods (on the left) and semi-spherical (on the right) type morphologies.
PMMA/PU (50/50)/CePO₄ Nanorods (0.1 wt.%)

PMMA/PU (60/40)/CePO₄ Nanorods (0.1 wt.%)

PMMA/PU (70/30)/CePO₄ Nanorods (0.1 wt.%)

PMMA/PU (80/20)/CePO₄ Nanorods (0.1 wt.%)

Fig. 6. CLSM images of CePO₄ nanorods in 0.1 wt.% into 50/50, 60/40, 70/30, and 80/20 ratio of PMMA/PU IPNs.
PMMA/PU (60/40)/CePO₄ Semi-spherical (0.1 wt.%)

PMMA/PU (70/30)/CePO₄ Semi-spherical (0.1 wt.%)

PMMA/PU (80/20)/CePO₄ Semi-spherical (0.1 wt.%)

Fig. 7. CLSM images of CePO₄ semi-spherical in 0.1 wt.% into 60/40, 70/30, and 80/20 ratio of PMMA/PU IPNs.

Fig. 8. TEM images of nanorods and semi-spherical CePO₄ nanoparticles dispersed in PMMA/PU 50/50 system.
2.1.1. X-Ray photoelectron spectroscopy (XPS) test was assessed in an Alpha 110, ThermoFisher Scientific XPS. XPS spectra of CePO₄ powders at two different pH are shown in Fig. 1. XPS spectra show typical binding energies of Ce³⁺ and Ce⁴⁺, which are forming the chemical state of as-prepared powders. The results were used to confirm the chemical composition.

2.2. Sequential synthesis of PMMA/PU/IPNs and CePO₄/PMMA/PU/IPNs

IPNs were prepared following the procedure in Ref. [1]. The transparency of the different ratios of PMMA/PU was evaluated by visual examination of samples synthesized, as it is shown in Fig. 2. Thermal properties of selected PMMA/PU pure samples were evaluated in Differential scanning calorimetry (DSC), conducted using a simultaneous Labsys Evo, Setaram TGA/DSC. Samples of DSC were tested at a heating rate of 10 K/min over the temperature range from 30 to 250 °C, under nitrogen atmosphere. Approximately 10–20 mg of each sample was placed in aluminium crucibles and maintained at 30 °C for 2 min, heated from 30 °C to 250 °C, maintained again at 250 °C for 2 min,
Fig. 10. Images of selected PMMA/PU/CePO₄ IPNs before (on the right) and after (on the left) aging during 500 h under heat and humidity conditions.
and cooled from 250 °C to 50 °C. PMMA/PU. Fig. 3 shows the thermograms corresponding to 60/40, 70/30, and 80/20 ratios of PMMA/PU.

Different amounts (0.1, 0.5, and 1.0 wt.%) of both nanostructures were incorporated and sonicated in the raw materials for PMMA and PU using two mixing times: 3 h and 10 min. Dibutyltin dilaurate (DBTDL) catalyst was added into the solution and sonicated for 10 min (Fig. 4). The final solution was poured into a PTFE mould and kept for 18 h and 3 h in UV lamp (363 nm).

Selected sample, PMMA/PU 50/50 ratio, is displayed in Fig. 5 showing the differences for each addition of nanorod and semi-spherical particles, respectively.

Dispersion of CePO₄ nanostructures in rod and semi-spherical morphologies within the PMMA/PU IPNs was performed by confocal laser scanning microscopy (CLSM) using a Carl Zeiss microscope (Carl Zeiss, Jena, Germany). The fluorescence intensity measurements were performed using the built-in software ZEN of the LSM 710. Results are shown in Figs. 6 and 7.

Transmission electron microscopy (TEM) was used to observe the dispersion of nanostructures in selected PMMA/PU/CePO₄ (50/50/1 wt.%). Ultra-thin sections were micromachined in epoxy resin. The sample was embedded and cured in R1078 Agar low viscosity resin (Agar Scientific). The sample was shaped into a pyramidal shape and cut with a diamond knife into slices. A slice was put onto a Holey Carbon film 200 mesh Copper (HC200Cu). TEM observations were recorded in a JEM-2000 FX electron microscope (JEOL). Results are shown in Fig. 8.

The surface of PMMA/PU/CePO₄ IPNs in 80/20 and 50/50 ratio with 0.1, 0.5, and 1 wt.% of CePO₄ were analyzed in a were evaluated using an Olympus BX51 microscope. Results are shown in Fig. 9.

Selected dog-bone specimen of PMMA/PU/CePO₄ were subjected to aging under humid conditions using a climate chamber (Memmer GmbH equipment) at 55 °C and 85% relative humidity for 500 h. Photographs of physical changes are shown in Fig. 10.

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